

Upper Atmosphere + Ionosphere Part-4

Diffusion : Molecular + Eddy:

Turbopause

Photo-ionization

Ionosphere : D, E, F1, F2 Regions

Ambipolar Diffusion: F2 Region

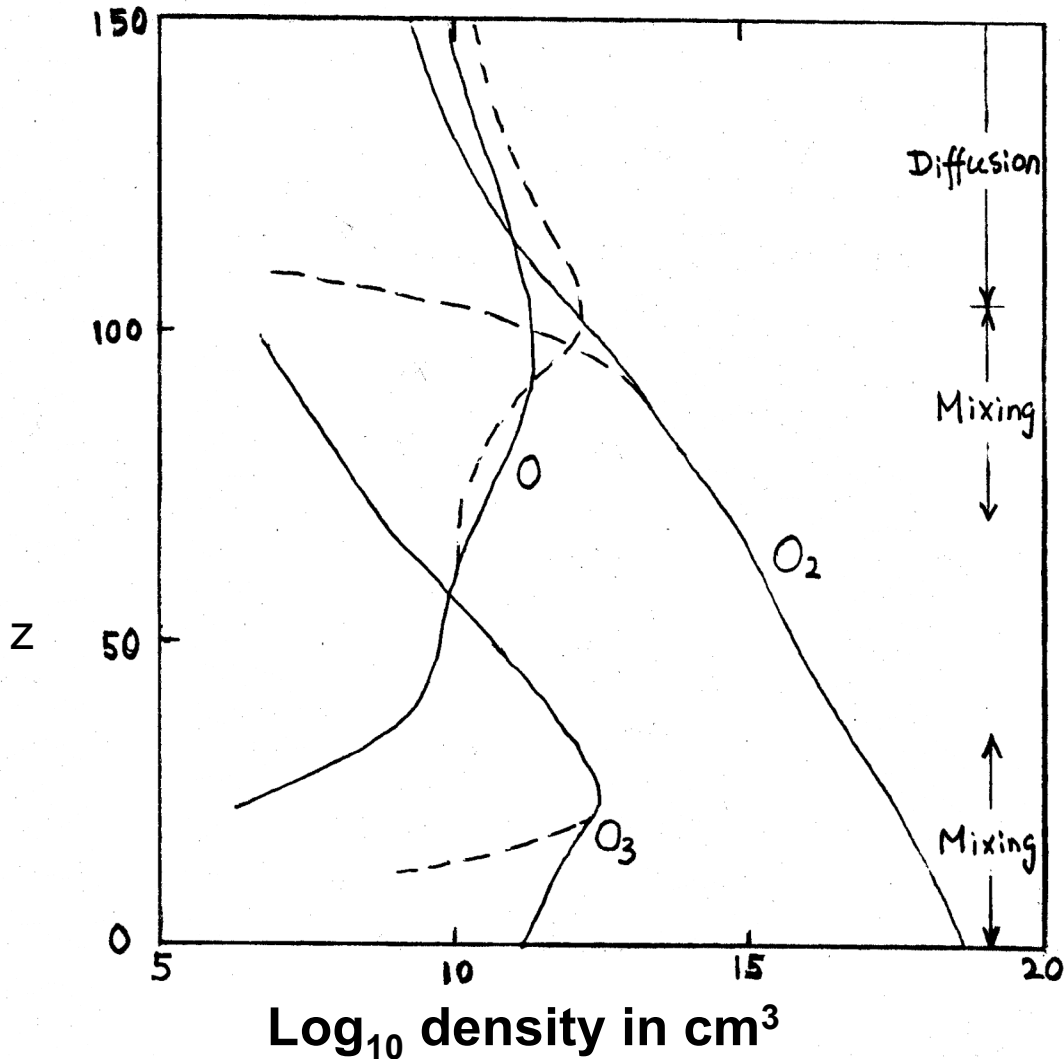
Radio Waves + the Ionosphere

Ionosphere of Other Planets

Oxygen Densities

Solid lines: mixing and diffusion:

Dashed lines: photochemical equilibrium (i.e. no mixing and diffusion)



NEED DIFFUSION

Continuity Equation with Diffusion

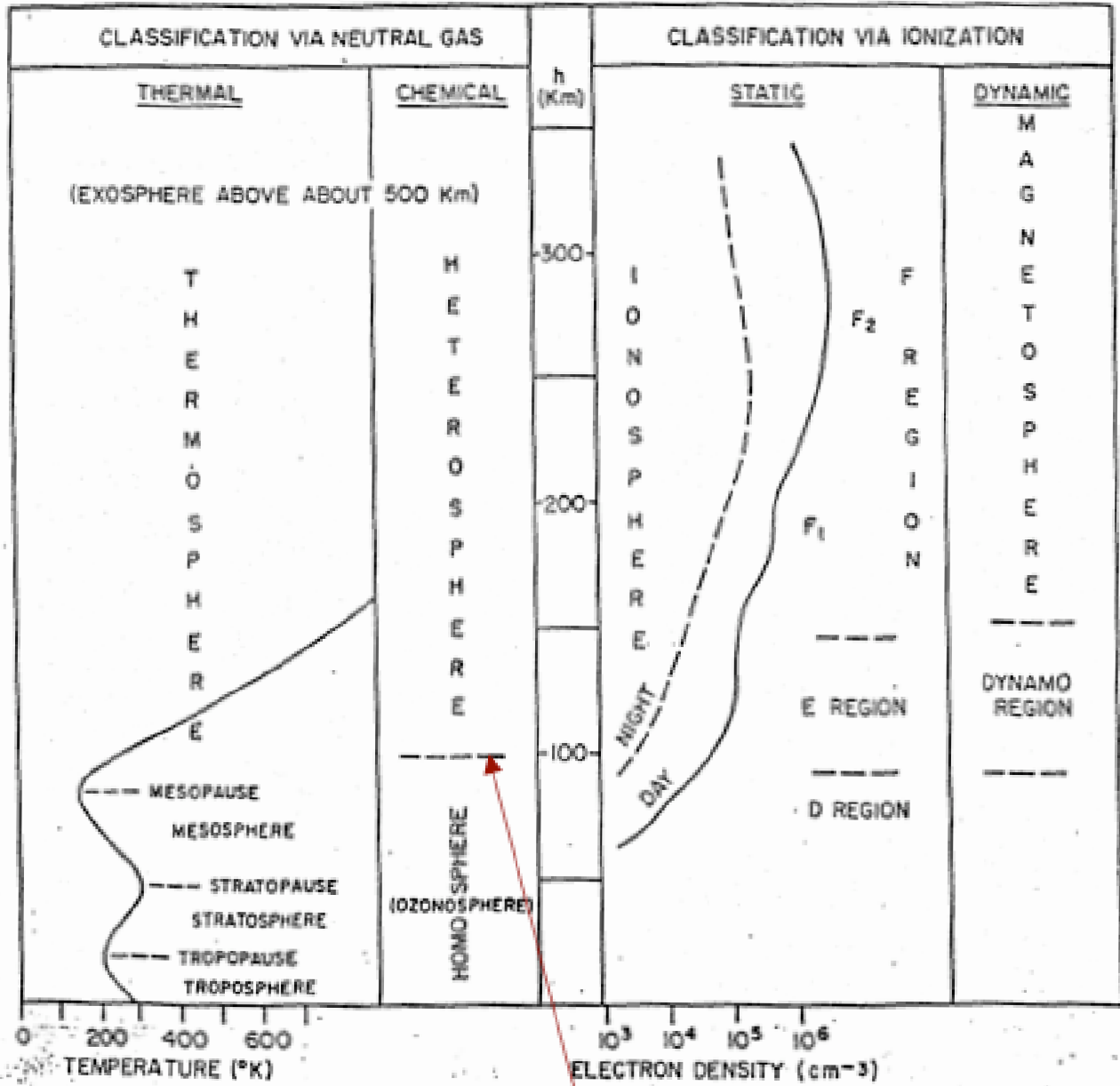
Rate of change of the number of species in dz

= Diffusive Flux + Prod.- Loss

$$\frac{dn_i}{dt} = \frac{d}{dz} \left[D_i \frac{dn_i}{dz} \right] + P_i - L_i$$

D is called the diffusion constant

Atmospheric Regions



Turbopause
Mixing ~ Diffusion

Mixing/Diffusion

Turbopause (Homopause)

The altitude where

Turbulent Mixing \approx Molecular Diffusion

Estimate the height of the turbopause

Eddy Mixing (Turbulence)

ignore the details of the flow at some scale

$$L_d \sim \text{the scale of the turbulence} = L_{\text{eddy}}$$

$$\bar{V}_d = \bar{V}_{\text{eddy}}$$

can create a diffusion coefficient, like a thermal conductivity

$$D_{\text{eddy}} \sim L_{\text{eddy}} \bar{V}_{\text{eddy}} ; L_{\text{eddy}} \sim \text{km}; \bar{V}_{\text{eddy}} \sim \text{m/s}$$

Molecular Diffusion (Turbulence)

Length scale is the distance between collisions, λ_{col}

$$D_d \sim \lambda_{\text{col}} \bar{v}_T, \text{ where } \bar{v}_T \text{ is the thermal speed}$$

$$\lambda_{\text{col}} \sim \bar{v}_T / [n\sigma_{\text{col}}]; \bar{v}_T \sim 600\text{m/s}; \sigma_{\text{col}} \sim 10^{-15}\text{cm}^2; n(0) = 2.5 \times 10^{19}/\text{cc}$$

$$\text{Set them equal; } D_d \sim D_{\text{eddy}}$$

and find turbopause; $n(z) \sim 10^{13}$ molecules/cc; or $z \sim 60\text{km}$

actually $\sim 100\text{km}$

Photochemistry and the Ionosphere

Simple Model



$$\frac{dn_{AB^+}}{dt} = J_i n_{AB} - \alpha n_e n_{AB^+}$$

$$\frac{dn_{AB^+}}{dt} = 0 \text{ in steady state}$$

$$\text{assume } n_e = n_{AB^+} \quad \text{charge neutrality}$$

get

$$n_{AB^+} \approx [n_{AB} J_i / \alpha]^{1/2}$$

Since $n_{AB} \propto \exp[-z/H_{AB}]$; $J_i \propto \exp[-\tau_i(z)]$

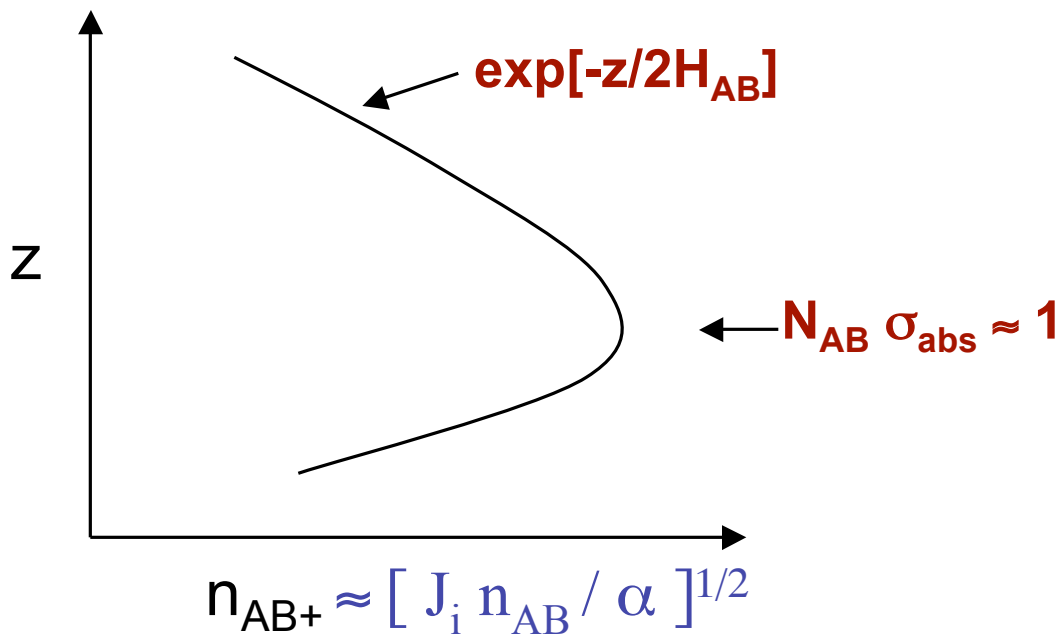
A Chapman Layer for Ion AB^+ is Formed

Ionospheres (continued)

Ionospheric layers are Chapman Layers

$$J_i = J_i^o e^{-\tau_i(z)}$$

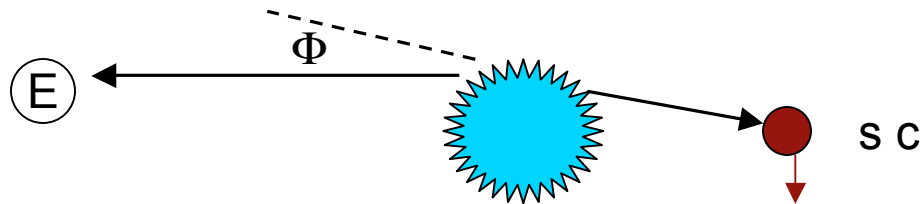
$$n_{AB} \approx n_{AB}(0) e^{-z/H_{AB}}$$



At top: ion scale height twice neutral H

RADIO WAVES

Spacecraft Observations of Ionosphere



**Measure changing index of refraction,
turn it into electron density vs z: $n_e(z)$
Then convert to neutral density using model**

(not required)

Compare phase of X band 8400 MHz; $\sim 3\text{cm}$
with the S band 2300 MHz; $\sim 10\text{cm}$

$$(*) \quad \Phi \approx - \frac{\omega}{c} \int_{\text{path}} (n-1) ds, \quad \omega = 2\pi\nu$$

$n = \text{index of refraction}$

Solve : $m_e \ddot{\bar{x}} = -m_e \nu_c \dot{\bar{x}} - e \bar{E}_0 e^{-i\omega t}$

steady state } $\bar{x} = \frac{1}{\omega + i\nu_c} \frac{e}{\omega m_e} \bar{E}_0 e^{-i\omega t}$
solution }

$\nu_c \ll \omega/2\pi$

$$n = \left[1 - \frac{\omega_p^2}{\omega^2} \right]^{1/2} \approx 1 - \frac{\omega_p^2}{2\omega^2}$$

$$\omega_p = \left[\frac{4\pi n_e e^2}{m_e} \right]^{1/2}$$

Using (*) $\Phi \propto \int_{\text{path}} n_e ds$

Ions are heavy so are much slower to respond:

Measure ionospheric density: $n_e \approx n_i \xrightarrow{\sim} [n_{AB}]^{1/2}$

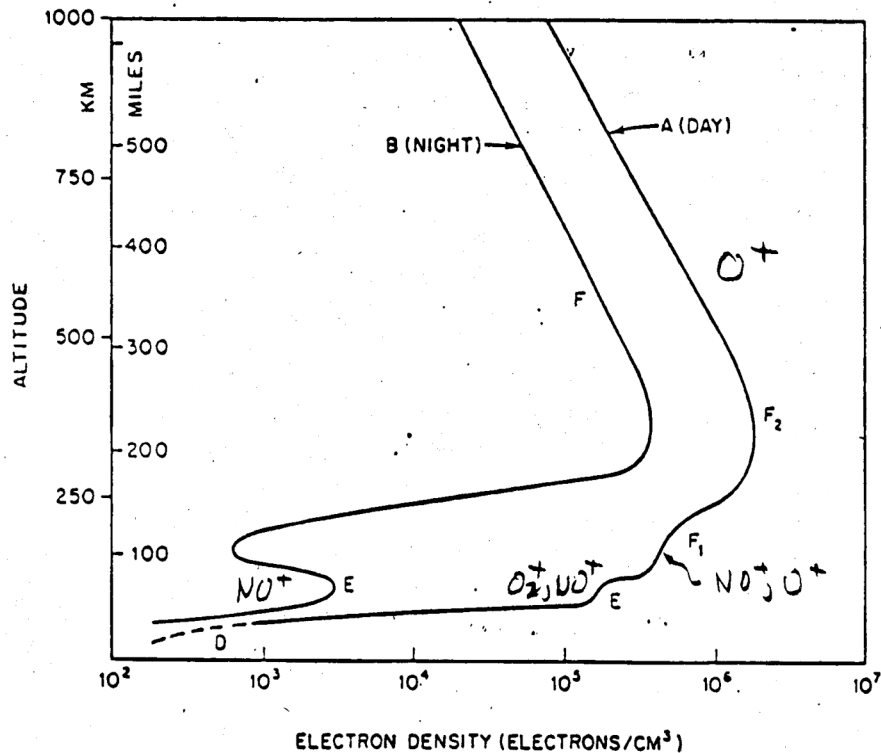
Result applies to most regions

(not F2 region where ambipolar diffusion determines scale height)

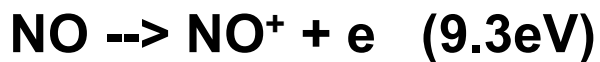
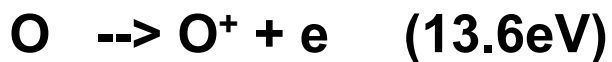
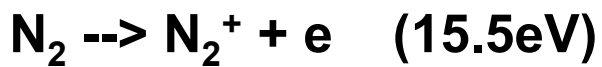
Earth's Ionospheric Layers

Day and Night

Principal Species



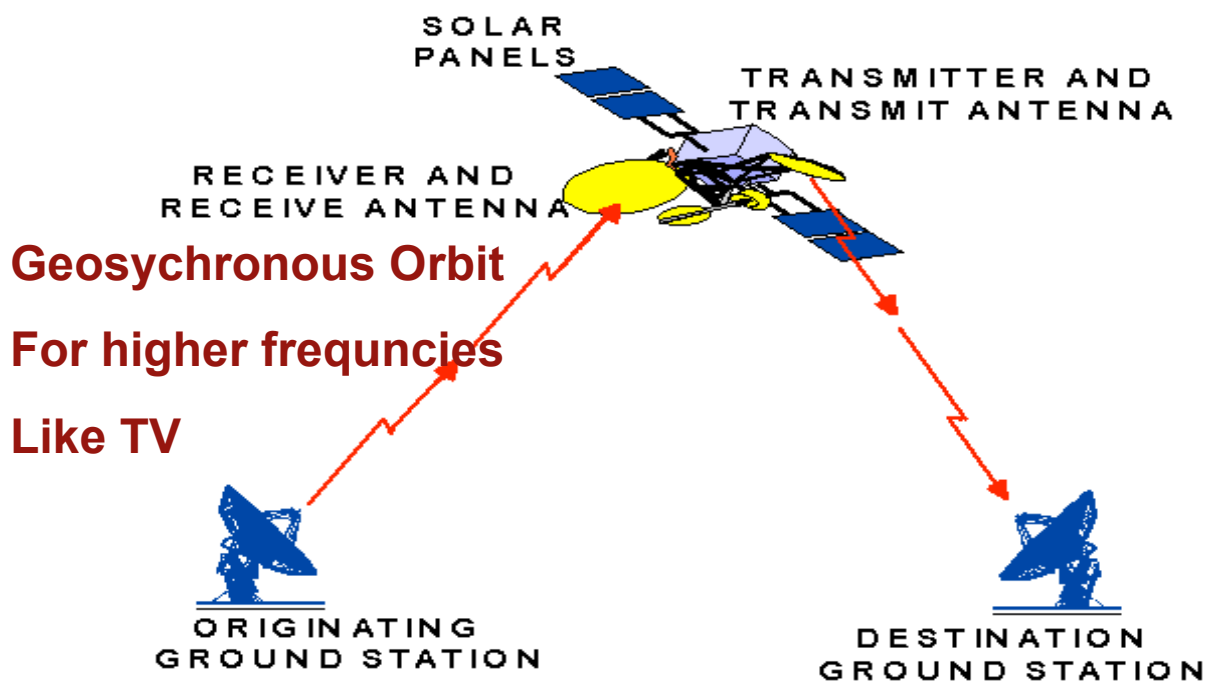
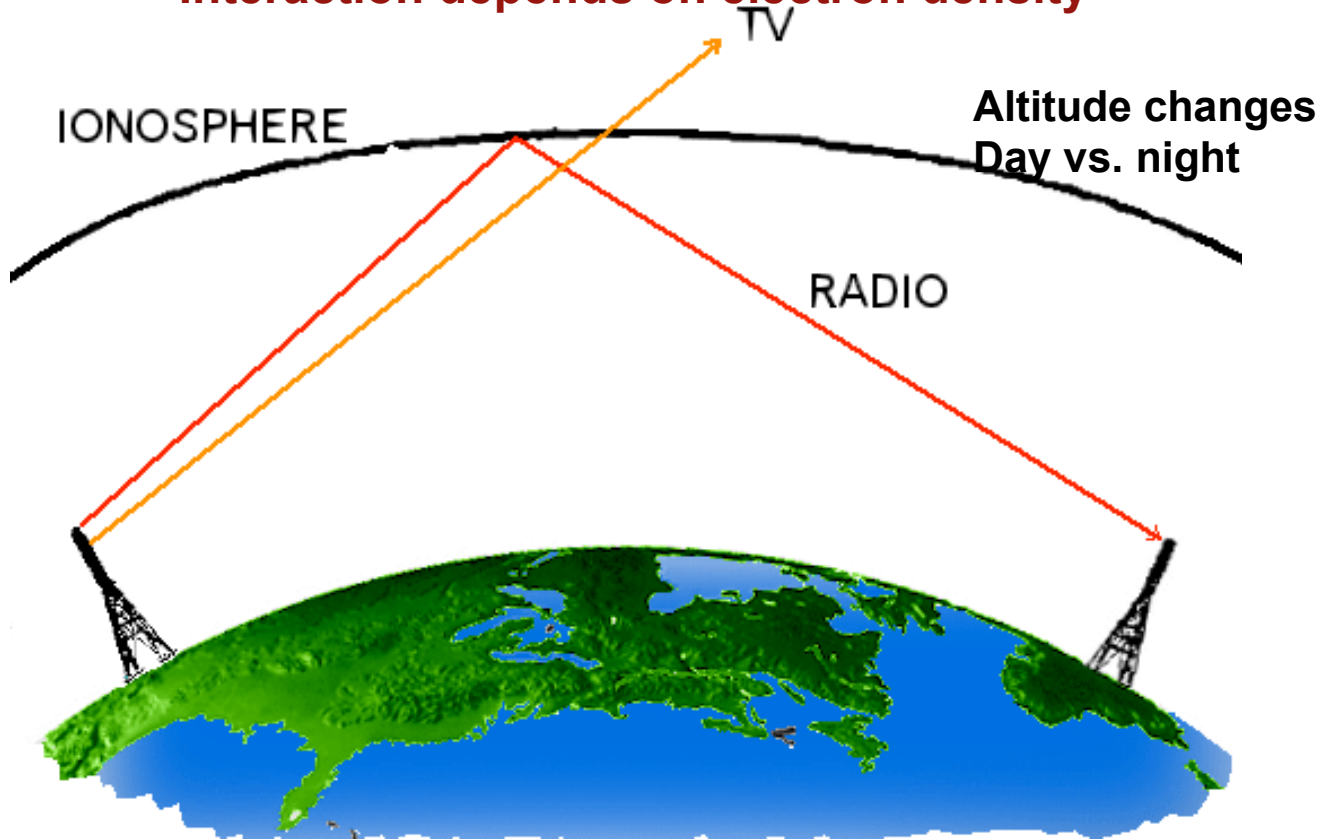
Favored ions have lowest ionization energies



Therefore, if made NO^+ is lives longer

Radio Waves and the Ionosphere

Interaction depends on electron density



Ions Produced



But ionosphere has a large, sometimes dominant, amount of NO^+



But ionosphere is predominantly O_2^+

Dominant Ions Observed

Ion-molecule reactions are fast and tend toward species with lowest ionization state:

$$I_A > I_B,$$



Metal ions, if available, dominate: Na, Mg, etc.

Earth's Ionosphere (D-region) in mesosphere

Initial Ions



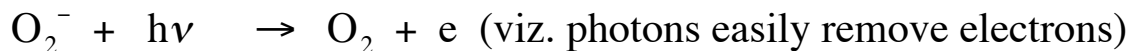
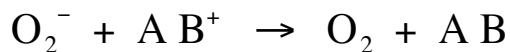
D region (relatively thick atmosphere, UV, viz. not efficient at ionizing)

~ 50 - 90 km

hard x-rays, lyman alpha photons, cosmic rays)

Weak, variable region, disappears rapidly at night

Density (n_M) large : negative are also formed



Ions $\text{O}_2^-, \text{O}_2^+, \text{H}_3\text{O}^+, \text{H}_5\text{O}_2^+$

$\text{CO}_3^-, \text{HCO}_3^-, \text{NO}_3^-, \text{CO}_3^-$

E-region \sim 110 km above mesopause

$$100 \text{ \AA} > \lambda > 30 \text{ \AA}, \lambda \sim 1000 \text{ \AA}, \lambda \sim 910 \text{ \AA}$$

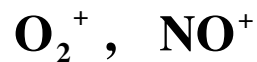
$$\text{\AA} = 10^{-8} \text{ cm} = 10^{-4} \mu\text{m}$$

Make N_2^+ , O_2^+ , O^+

Then



Observe



Loss : Electron - Ion Recombination



Note : Meteor Showers also add metal ions
and the presence of NO neutrals



enhances NO^+ density

E Region (cont)

Time Constant : E – Region

Look only at reactions 3 + 5

$$\frac{dn_{\text{NO}^+}}{dt} \approx k_1 n_{\text{O}^+} n_{\text{N}_2} - \alpha_{\text{NO}^+} n_{\text{NO}^+} n_e$$

$$n_e = n_{\text{NO}^+} + n_{\text{O}_2^+} \approx n_{\text{NO}^+} \quad : \text{ most are NO}^+$$

Time constant : $t_{\text{NO}^+} \sim n_{\text{NO}^+} / [dn_{\text{NO}^+}/dt]$:

$$\sim \frac{n_{\text{NO}^+}}{k_1 n_{\text{O}^+} n_{\text{N}_2}} \quad \text{or} \quad \frac{n_{\text{NO}^+}}{\alpha_{\text{NO}^+} n_{\text{NO}^+}^2}$$

$$t_{\text{NO}^+} \sim \frac{1}{\alpha_{\text{NO}^+} n_{\text{NO}^+}} ; \alpha_{\text{NO}^+} \approx 3 \times 10^{-7} \text{ cm}^3/\text{s}$$

Day $n_{\text{NO}^+} \approx 3 \times 10^5 / \text{cm}^3$

$$t_{\text{NO}^+} \approx 10 \text{ sec}$$

Night $n_{\text{NO}^+} \approx 3 \times 10^3$

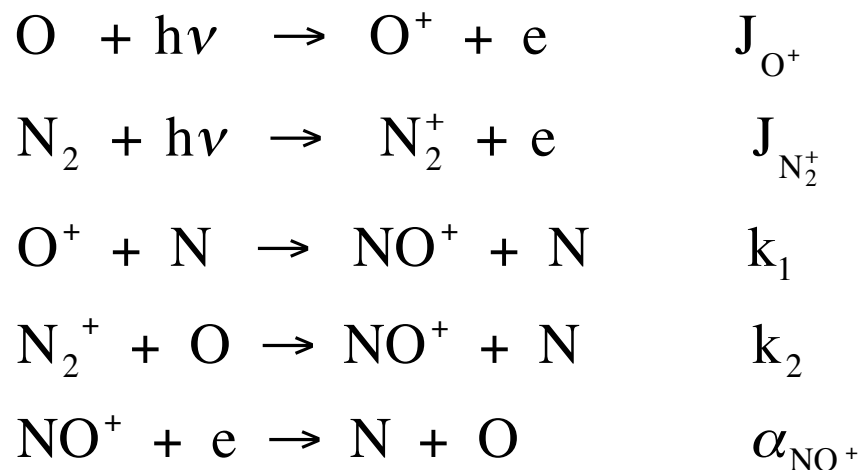
$$t_{\text{NO}^+} \approx 10^3 \text{ sec} \rightarrow 15 \text{ hours}$$

Decays but does not go away at night

F1 Region (~180km)

F₁ Region

O and N₂ dominate (O₂ has a smaller scale height)



O⁺ and NO⁺ are the observed ions

F₁ Region (cont.)

O⁺ Equations

Form O⁺ - Lose O⁺

$$0 = \frac{dn_{O^+}}{dt} = J_{O^+} n_O - k_1 n_{O^+} n_{N_2}$$

$$n_{O^+} = \frac{J_{O^+} n_O}{k_1 n_{N_2}}$$

NO⁺ Equations

Form NO⁺ - Lose NO⁺

$$0 = \frac{dn_{NO^+}}{dt} = k_1 n_{O^+} n_{N_2} - \alpha_{NO} n_{NO^+} n_e$$

Combine Eq.

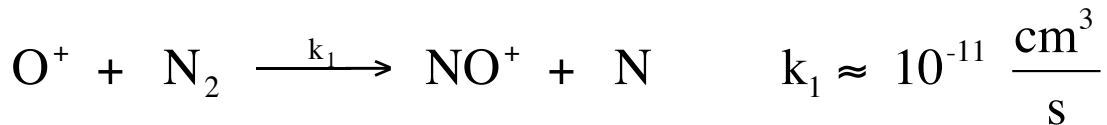
$$0 = J_{O^+} n_O - \alpha_{NO^+} n_e^2$$

$$n_e = \sqrt{\frac{J_{O^+} n_O}{\alpha_{NO^+}}}$$

like the simple first model: $n_e \propto n_O^{1/2}$

TIME CONSTANTS: F1 Region

F₁ region $n_O > n_{N_2} \gg n_{O_2}$



$$\frac{dn_{O^+}}{dt} = J_{O^+} n_O - k_1 n_{O^+} n_{N_2} = 0$$

$$\text{Solve : } n_{O^+} = \frac{J_{O^+} n_O}{k_1 n_{N_2}}$$

$$\text{Time constant : } t_{O^+} = \frac{1}{k_1 n_{N_2}}$$

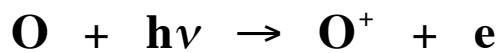
$$200\text{km} \quad n_{N_2} \approx 10^{10} / \text{cm}^3$$

$$\text{Time for } O^+ = 100 \text{ sec}$$

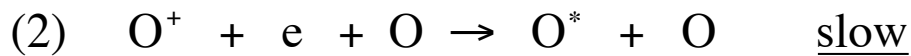
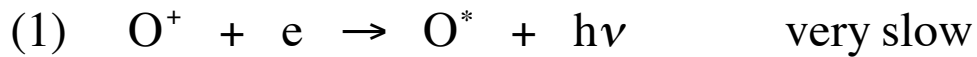
Mostly disappears at night

F_2 – region > 250 km

Source (O is dominant neutral)



But



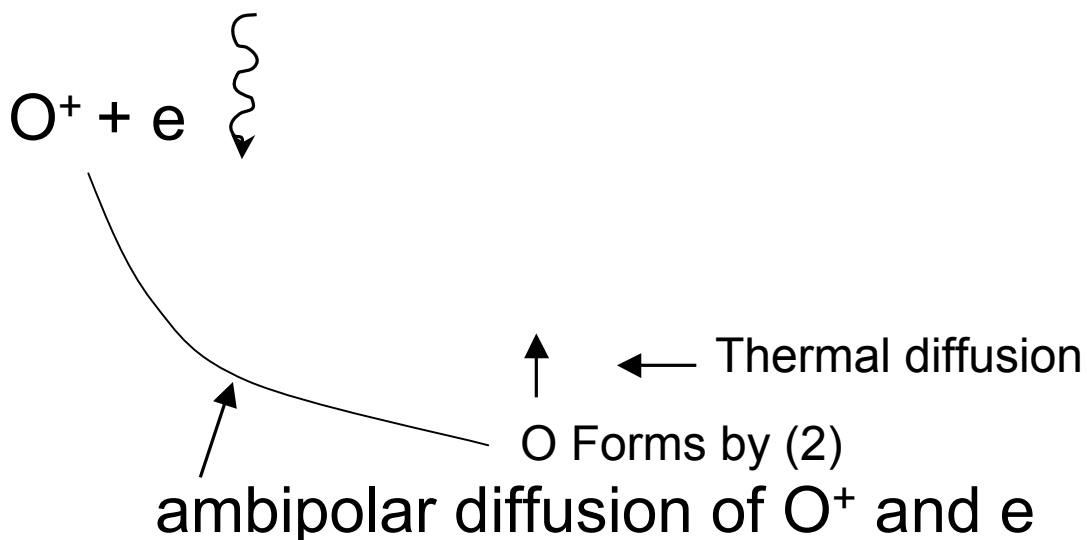
Like thermospheric heat

O^+ diffuses down to where loss [(2)] is fast

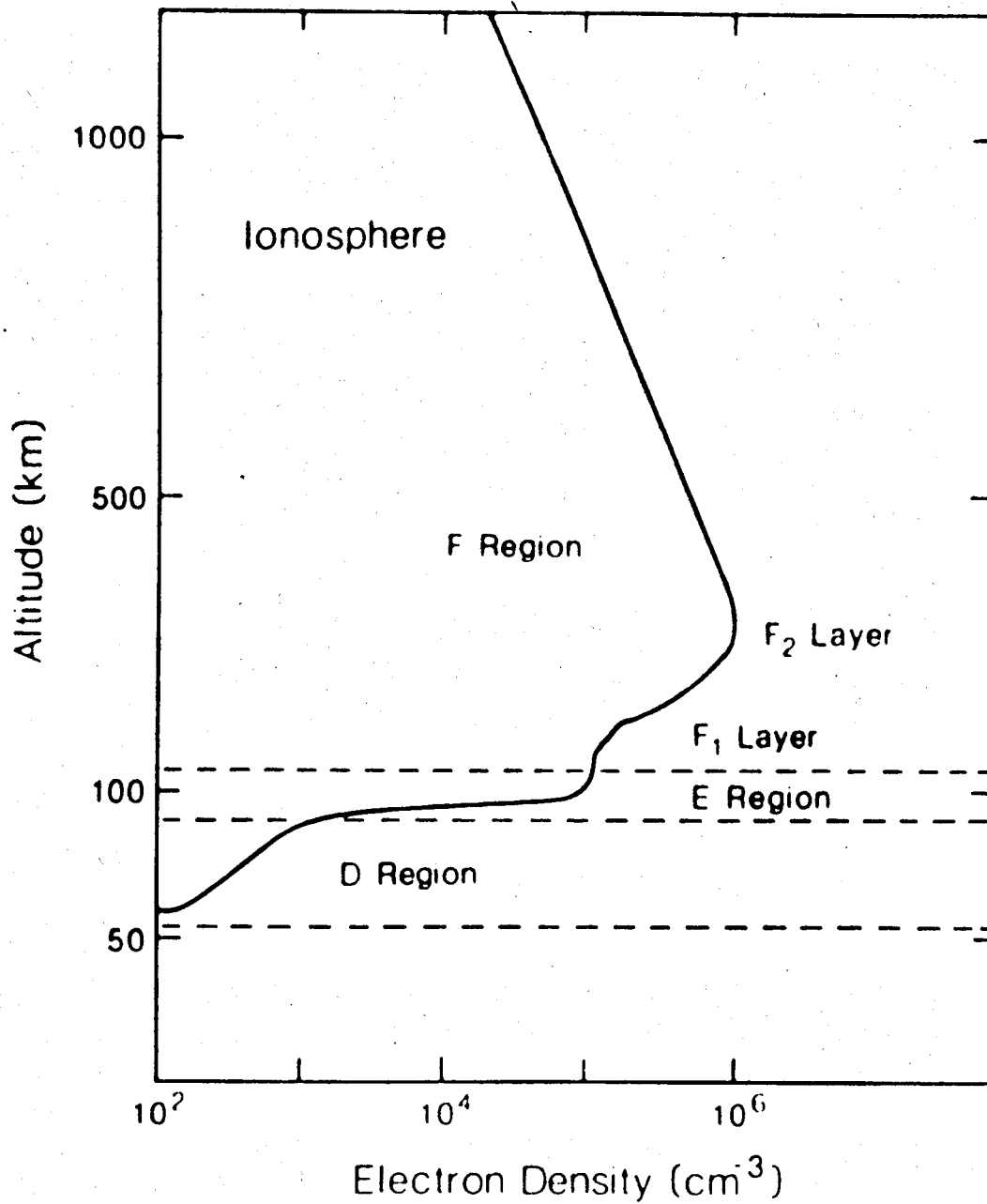
BUT

If charges diffuse \Rightarrow currents!

Ambipolar Diffusion



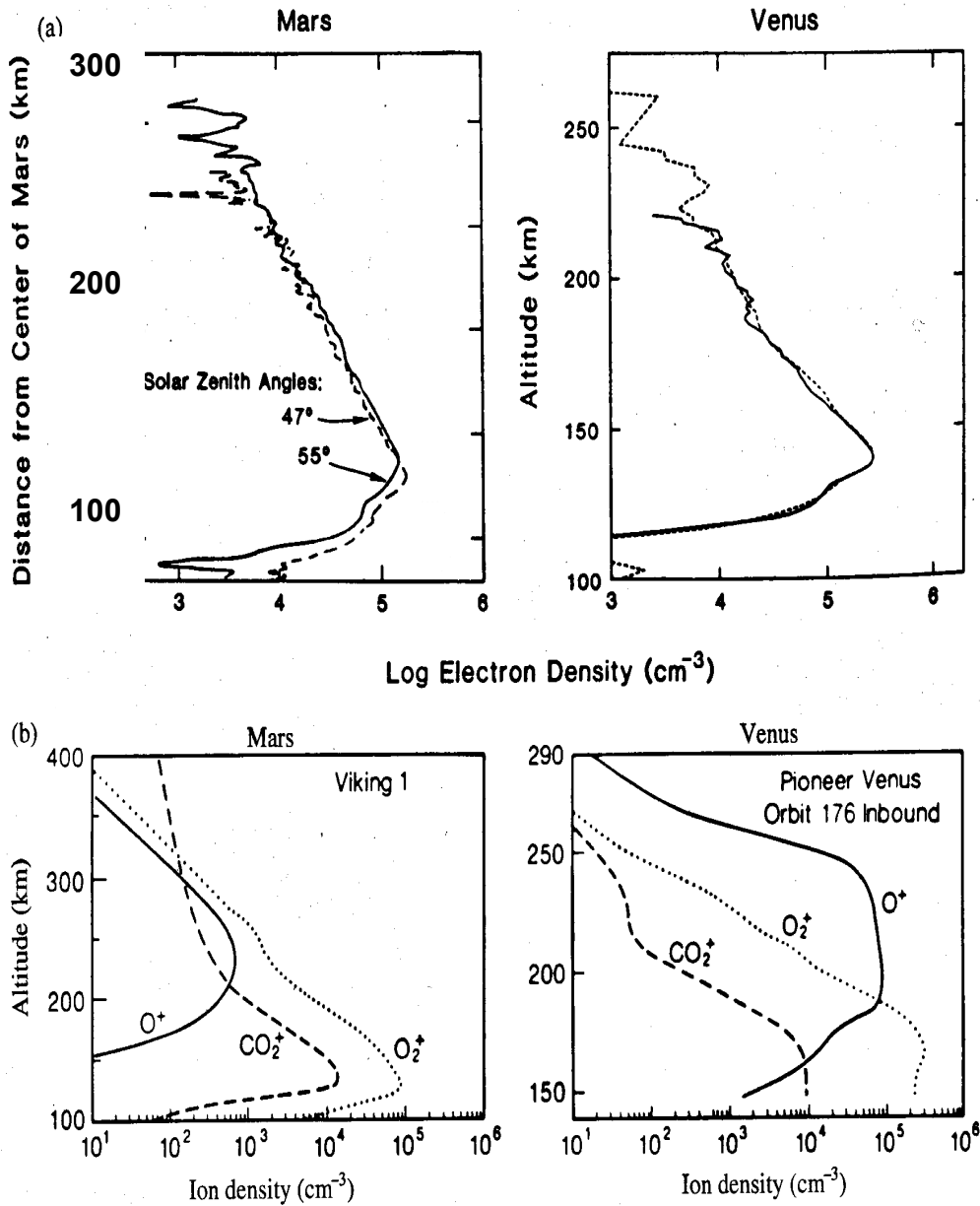
Earth's Ionospheric Layers Day Time



Mars and Venus Ionospheric Layers

Chapman Layers

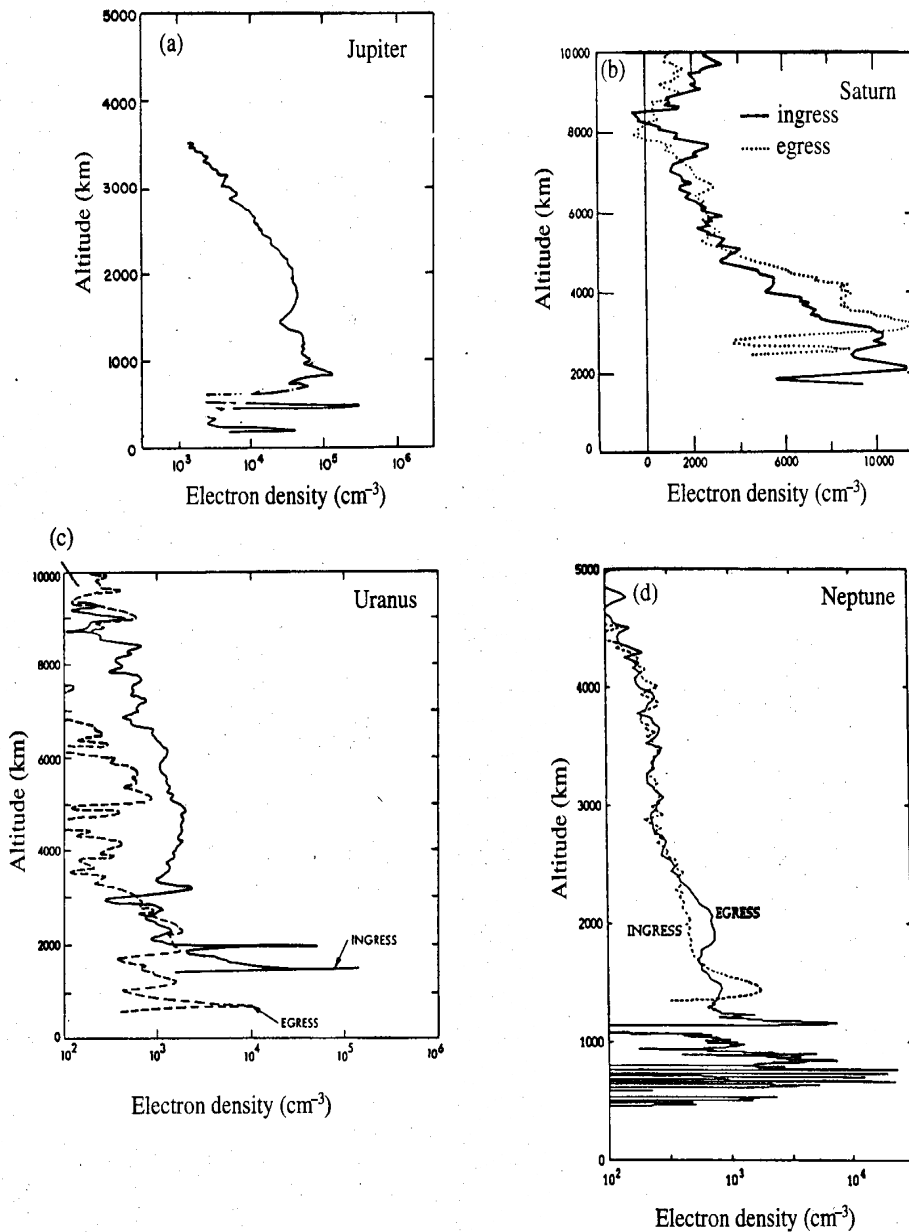
Radio Occultation Measurements



**At top interaction with the sun's
sields and plasma becomes important**

Giant Planet Ionospheres

Chapman structure + waves



Ingress and Egress apply to Voyager
radio wave results

Aurora

Interaction of the solar plasma and fields with the earth's upper atmosphere

Communications affected by solar activity



Mostly, but not exclusively, lines of O or O⁺

#4 Summary

Things you should know

Turbopause (Homopause)

Diffusive Term

Photo-ionization

Photochemistry of Ionosphere

Layered Ionosphere

Ambi-polar Diffusion

Refraction of SC Radio Signal

Ionospheres of Other Planets